PROJECT FOR TERM II

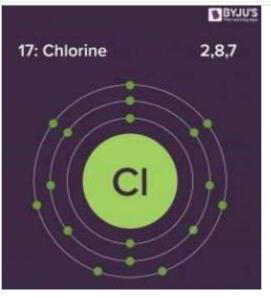
CLASS VIII

SCIENCE

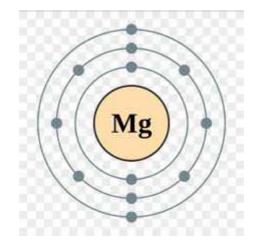
Question : PREPARE THE MODEL OF THE FOLLOWING ELEMENTS GIVEN BELOW. DIAGRAM OF ELECTRONIC CONFIGURATION ARE GIVEN FOR YOUR REFERENCE. (Any six, you can do either 3 dimensional structure or the simple one) LAST DATE OF SHOWING: 19.12.2020

Distribution of electron

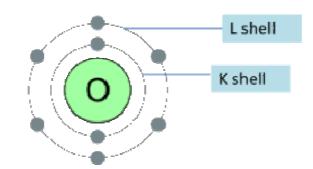
Shell and 'n' value	Max. Electrons in the Electron Configuration
K shell, n=1	$2^*1^2 = 2$
L shell, n=2	$2^{*}2^{2} = 8$
M shell, n=3	2*3 ² = 18
N shell, n=4	$2^{*}4^{2} = 32$



ELECTRONIC CONFIGURATION OF CHLORINE



ELECTRONIC CONFIGURATION OF MAGNESIUM

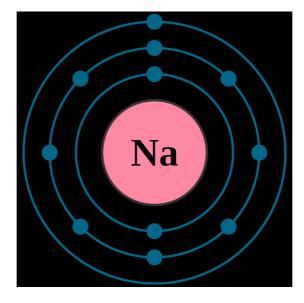


Oxygen (2,6) (8 electrons =1st and 2nd orbit)

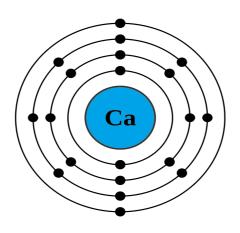
ELECTRONIC CONFIGURATION OF OXYGEN



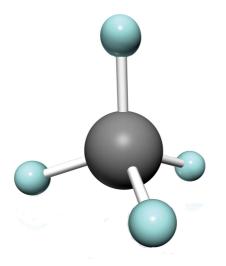
ELECTRONIC CONFIGURATION OF NITROGEN



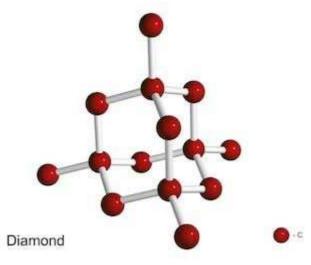
ELECTRONIC CONFIGURATION OF SODIUM



ELECTRONIC CONFIGURATION OF CALCIUM

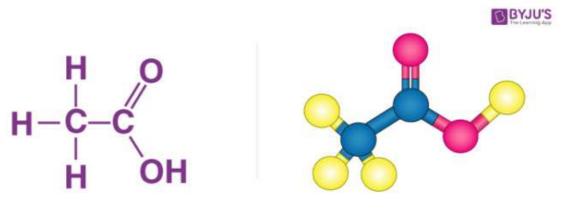


THREE DIMENSIONAL STRUCTURE OF METHANE (CH₄)



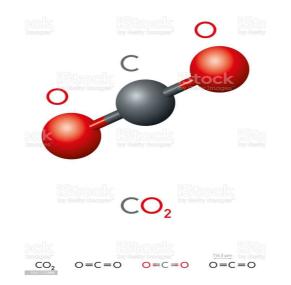
shutterstock.com · 24510613

THREE DIMENSIONAL STRUCTURE OF DIAMOND

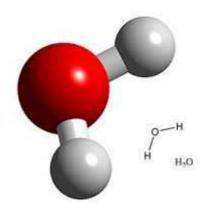


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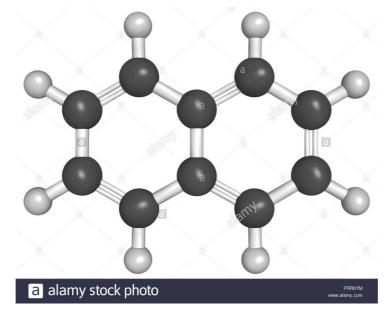
THREE DIMENSIONAL STRUCTURE OF ACETIC ACID



THREE DIMENSIONAL STRUCTURE OF CARBONDIOXIDE



THREE DIMENSIONAL STRUCTURE OF WATER



THREE DIMENSIONAL STRUCTURE OF NAPHTHALENE